



# University of Pretoria Yearbook 2016

## General chemistry 171 (CHM 171)

<b>Qualification</b>	Undergraduate
<b>Faculty</b>	<a href="#">Faculty of Natural and Agricultural Sciences</a>
<b>Module content</b>	General introduction to inorganic, analytical and physical chemistry. Nomenclature of inorganic ions and compounds, stoichiometric calculations concerning chemical reactions, redox reactions, solubilities and solutions, atomic structure, periodicity. Molecular structure and chemical bonding using the VSEPR model. Principles of reactivity, electrochemistry, energy and chemical reactions, entropy and free energy. Appropriate tutorial classes and practicals.
<b>Module credits</b>	16.00
<b>Programmes</b>	<a href="#">BEng Chemical Engineering</a> <a href="#">BEng Chemical Engineering Engage</a> <a href="#">BEng Civil Engineering</a> <a href="#">BEng Civil Engineering Engage</a> <a href="#">BEng Electrical Engineering</a> <a href="#">BEng Electronic Engineering</a> <a href="#">BEng Industrial Engineering Engage</a> <a href="#">BEng Metallurgical Engineering</a> <a href="#">BEng Mining Engineering</a>
<b>Service modules</b>	Faculty of Engineering, Built Environment and Information Technology
<b>Prerequisites</b>	No prerequisites.
<b>Contact time</b>	1 web-based period per week, 1 practical per week, 1 discussion class per week, 4 lectures per week
<b>Language of tuition</b>	Both Afr and Eng
<b>Academic organisation</b>	Chemistry
<b>Period of presentation</b>	Semester 1

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